Antioxidant Capacity, Oxidative Stability and Sensory Properties of Flavored Butter Using Dried Apricot Pulp

M. Bulut¹, and E. Gundogdu^{1*}

ABSTRACT

In this study, various concentrations of Dried Apricot Pulp (DAP) were added to butter at levels of 0, 15, 20, and 25%, and its effects on antioxidant capacity, oxidative stability, color, and sensory attributes were assessed during 45 days of storage. Butter samples were labeled as A_0 (Control), A_{15} , A_{20} , and A_{25} corresponding to the respective levels of DAP incorporation. The addition of DAP resulted in significant reductions in pH, while titratable acidity (expressed as lactic acid %) and free fatty acids (FFA) content (mg KOH
g⁻¹ butter) increased in a dose-dependent manner (P< 0.05). Fortification with DAP delayed peroxide formation in the butter samples. Additionally, DAP supplementation significantly (P< 0.05) increased Total Phenolic Content (TPC), improved DPPH (2,2- Diphenyl-1-Picrylhydrazyl) scavenging activity, and increased Total Antioxidant Capacity (TAC) during storage. Panelists assigned higher sensory scores to the DAPenhanced butters. These fortified butters exhibited lower fat content (ranging from 65.25 to 65.50%) compared to the control butter (73.75%). Notably, the total sugar content of samples A_{15} , A_{20} , and A_{25} was 3.69±0.06, 4.70±0.06, and 5.64±0.08, respectively. Overall, this study demonstrates the potential of all DAP ratios, particularly 25%, as rich sources of antioxidants. However, further formulation adjustments and comprehensive analyses are warranted for industrial and marketing applications.

Keywords: Enhancing butter, Fortified butter, Stability, Total sugar content.

INTRODUCTION

Butter is a water-in-oil emulsion containing approximately 79–84% lipid content and enriched with vitamins A, E, D, and K (Ebrahimian et al., 2023). It is widely utilized as a spread on plain or toasted bread products, a condiment on cooked vegetables, and an ingredient in various culinary applications, including baking, sauce preparation, and pan-frying (Bujancă et al., 2016). Butter products are typically categorized as sweet cream unsalted, salted sweet cream, cultured unsalted, cultured salted butter, or traditional sour cream butter (Kwak et al., 2013). Additionally, the Turkish Food Codex classifies butter as yayık butter, flavored butter mix, and flavored butter produced with various spices, vegetables, honey, other foodstuffs, and fruits (Notification No. 2005/19). Enhancing butter with natural ingredients that provide health benefits is one strategy to promote its consumption (Vidanagamage et al., 2016).

Numerous studies have focused on enhancing the nutritional properties of butter due to its widespread consumption globally (Thakaeng et al., 2020). These studies primarily aim to inhibit butter oxidation using plant derivatives (Bule et al., 2022; Çakmakçı et al., 2014; Dagdemir et al., 2009; Ebrahimian et al., 2023; Gramza-Michalowska et al., 2007; Wojdyło et al., 2005; Ziarno et al., 2023).Although plant extracts have been

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effective in enhancing antioxidative stability, they have also been reported to compromise the sensory properties of butter. For instance, Nadeem et al. (2013) found that Moringa oleifera leaf extract negatively affected the taste and aroma of butter compared to the control butter. Similarly, Göksel Saraç and Dogan (2016) demonstrated that the addition of dietary fibers from vegetable and fruit wastes, such as stone pears, celery root, celery leaves, spinach, and orange, reduced butter's sensory scores. Sensory characteristics significantly influence consumer preference (Markey et al., 2017). Therefore, fortification should not only enhance nutritional value but also improve consumer appeal (Alqahtani et al., 2023). Zakharova (2014) reported that vegetable fats, proteins, natural vegetable fillers, and fruits were used in the dairy industry to maintain traditional nutritional value and compensate for nutrient deficiencies. Consequently, fortifying dairy products with fruit can enhance functional properties, improve appearance, and add color (Salehi, 2021).

Among various fruits, apricot (Prunus armeniaca L.) is particularly popular due to its pleasant taste, unique flavor, and appealing vivid color, whether consumed fresh or dried (Ali et al., 2011). Dried apricot is especially noted for being a rich source of phenolics, vitamin C, and carotenoids (Hussain et al., 2013). Fructose and sucrose are the dominant sugars in apricots, contributing to their sweetness, sensory quality, and consumer satisfaction (Fan et al., 2017).

Despite extensive research on fruitfortified dairy products like yogurt, ice cream, kefir, and cheese, to our knowledge, no studies have investigated butter produced with apricot pulp. This study aimed to improve the antioxidant properties of flavored butter with dried apricot pulp (DAP) to enhance its nutritional value, assess the impact of DAP on chemical properties, enhance sensory properties such as taste and color, and develop an alternative butter variety comparable to other dairy products.

MATERIALS AND METHODS

Materials

The cream (65% milk fat) was procured from Izi Milk and Milk Product Co. (Konya, Turkey), and dried apricot was sourced from a local market in Gümüşhane, Turkey. Chemicals of analytical grade were obtained from Merck Co. (Darmstadt, Germany) and Sigma Chemical Co. (St. Louis, MO).

Methods

Preparation of DAP

The dried apricots were washed and sliced. The slices were mixed with water (1:1) and boiled for 30 minutes until softened. The mixture was then blended into a pulp using an electric blender (Waring Commercial, 38BL40, USA) and stored in a sterilized glass bottle at 4°C until used for butter production.

Preparation of Butter Samples

The cream, adjusted to 50% fat with skim milk, was churned, washed, and divided into four parts. Apricot pulp was added to the butter samples at concentrations of 15, 20, and 25% (w/w), coded as A_{15} , A_{20} , and A_{25} , respectively. The fourth sample, without any additives, served as the control (A_0) . The butter samples were packaged with stretch film and covered with aluminum foil to protect them from light, then, stored at 4°C for two months. All analyses were conducted on days 1, 15, 30, 45, and 60, except for proximate analysis, which was performed only on the first day. The first sampling was done one day after storage, and all butter samples were produced in duplicate.

Physicochemical Properties of DAP

The dry matter (at 105°C) and ash content (at 550°C) of DAP were determined using the methods described by Khairuddin et al. (2017). The pH was measured with a pH meter (HANNA Instruments, Italy), and color values (L^*, a^*, b^*) were assessed using a Minolta Chromameter (CR-200) (Konica Minolta Sensing, Inc. Japan). Color values were: L (brightness), $100=$ White, $0=$ Black; a (+) red, a (-) green; b (+) yellow, b (-) blue.

Proximate and Physicochemical Properties of Butters

Dry matter, fat content, and sugar content were determined on the first day of storage, while pH, titratable acidity, free fatty acids, and color values were determined on days 1, 15, 30, 45, and 60 of storage. Dry matter (oven drying), fat content (Gerber method), the pH of butter samples was measured using a pH meter (HANNA Instruments, Italy), and color values (L^*, a^*, b^*) were determined using a Minolta colorimeter (Chroma Meter, CR-200, Osaka, Japan). Titratable acidity, expressed as lactic acid percentage, and Free Fatty Acid (FFA) content, expressed as mg KOH g^{-1} butter, were determined using the method described by Timtey et al. (2024) on days 1, 15, 30, 45, and 60 of storage.

Determination of Sugar Content

Fructose, glucose, and total sugar analyses were conducted on both DAP and butter samples on the first day of storage. The procedure of the International Honey Commission (Anonymous, 2009) was used for sugar analysis. The analysis was performed using HPLC-RID (Thermo Scientific Products Finnigan Spectra System). Sugar separation was achieved with a column (5 μm particle size, L×ID 250×3 mm). The flow rate was 1.3 mL min⁻¹, and acetonitrile/water (80:20, v/v) was used as the mobile phase. The column temperature was maintained at 50 ± 1 °C, and the injection volume was 20 µL.

Determination of Peroxide Value

Peroxide value (PV) was determined to assess primary oxidation during storage. Samples were dissolved in a mixture of chloroform and glacial acetic acid and titrated with 0.002 N $Na₂S₂O₃$ using a 1% starch indicator. PV was expressed as meq O_2 kg⁻¹ butter (Yatsenko *et al.*, 2020).

Determination of TPC, DPPH Radical Scavenging Capacity, and TAC

First, the extract was prepared. A sample $(1 \text{ g}, \text{ pulp butter}^{-1}), 2 \text{ mL of hot distilled}$ water, and 5 mL of ethanol were extracted in an ultrasonic bath (Sonorex Super RK 103 H, Germany). After filtering through filter paper, the extract was passed through 0.45 μm filters and used for Total Phenolic Content (TPC) , DPPH, and TAC tests.

 TPC was determined using the Folin– Ciocalteu test. To calibrate the concentration of total phenolics in butter samples, the absorbance of standard Gallic acid solutions (20, 40, 60, 80, 120, and 160 μ g mL⁻¹) was measured at 760 nm (Shimadzu UV-1800 240V, Japan) , and results were expressed as mg Gallic Acid Equivalent (GAE) kg⁻¹ sample (Kasangana et al., 2015).

For DPPH analysis , 100 μL of the extract was mixed with 3,000 μL of DPPH solution and vortexed. After a 30-minute incubation, absorbance was measured at 517 nm using a spectrophotometer (Shimadzu UV-1800 240V, Japan), with 100 μ L of methanol as the blank. DPPH radical scavenging activity was expressed as inhibition capacity percentage (Ahmed et al., 2015).

TAC was measured using the equation of the graph prepared with ascorbic acid solutions (25, 50, 100, 150, 250, 500, and 900 μ g mL⁻¹), and TAC was expressed as

mg Ascorbic Acid Equivalent (AAE) kg⁻¹ butter. Then, 500 µL of the extract was mixed with 2500 µL of deionized water and $1,000 \mu L$ of molybdate reagent. The mixture was vortexed and incubated at 95°C for 90 minutes, then, cooled to room temperature. Absorbance was measured at 695 nm (Shimadzu UV-1800 240V, Japan), against a blank which contained distilled water instead of the sample (Kasangana et al., 2015).

Sensory Analyses

A panel of 14 trained specialists familiar with dairy products from the Food Department, Gümüshane Engineering University, Turkey, evaluated the butter samples. Approximately 15 g of each butter sample was placed in cups labeled with random three-digit numbers. All samples were allowed to reach room temperature for 30 minutes before being served to the panelists, who were given instructions prior to sensory analysis. Bread was provided for testing the butters, and water was available for rinsing the mouth (Zine-Eddine et al., 2022). Panelists evaluated the samples for colour, texture, odor, taste, rancidity and general acceptability. Sensorial attributes were scored according to Hedonic scale from 1 (lowest) to 9 (highest).

Statistical Analyses

ANOVA was employed to analyze the differences among the group means for all data. Mean values showing significant differences were compared using Duncan's multiple range tests ($P < 0.05$). The results were expressed as means of two replications. For this analysis, the SPSS 24.0 software package (SPSS Inc., Chicago, IL, USA) was used.

RESULTS AND DISCUSSION

Physicochemical Properties of DAP

Table 1 presents the physicochemical composition of DAP, highlighting its high Total Phenolic Content (TPC), DPPH radical scavenging activity, and Total Antioxidant Capacity (TAC). The literature reports varying results concerning the chemical composition, antioxidant properties, color, and phenolic content of dried apricot. Karabulut et al. (2018) found that dry matter, glucose, and fructose contents ranged between 84.1-89.3 and 18.26-13.48%, respectively. Canadanović-Brunet et al. (2013) reported a total phenolic content of 498.13±12.04 mg GAE 100 g⁻¹ in dried apricot. Ivanova et al. (2017) documented L, a^* , and b^* values as 30.2-

Table 1. Physicochemical properties of Dried Apricot Pulp used during butter production.^{*a*}

Properties	DAP		
Dry matter $(\%)$	46.07 ± 0.15		
Ash $(\%)$	1.60 ± 0.11		
pH	4.18 ± 0.01		
L^*	48.80 ± 0.55		
a^*	3.70 ± 0.37		
h^*	37.20 ± 0.56		
TPC $(GA \text{ mg kg}^{-1})$	1592.08±57.45		
DPPH (Inhibition %)	40.57 ± 0.00		
TAC $(AA \text{ mg g}^{-1})$	639.11 ± 0.94		
Fructose $(\%)$	10 ± 0.01		
Glucose $(\%)$	18 ± 0.01		
Total sugar $(\%)$	28 ± 0.01		

a Data are means±standard deviation.

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36.4, 6-12.4, and 9-21.8, respectively. These discrepancies can be attributed to different apricot varieties, geographic conditions, drying techniques and temperatures (sulphured vs. unsulphured), maturity stages, storage conditions, chemical compositions, and analytical methods.

Proximate and Physicochemical Properties of Butters

The chemical properties of butters produced with different concentrations of DAP (15, 20, and 25%) were compared with those of the control. Analyses of dry matter, fat, and sugar content were performed on the first day of storage, as detailed in Table 2. Additionally, pH, titratable acidity, Free Fatty Acids (FFA), and color properties were assessed throughout the storage period, as presented in Table 3.The addition of DAP had a statistically significant effect $(P< 0.05)$ on the chemical properties. Specifically, the addition of pulp decreased dry matter and fat content while increasing sugar content, depending on the pulp concentration. These findings align with those of Mohamed and Shalaby (2016), who reported that the addition of apricot pulp to processed cheese spread reduced dry matter content.

The pH values of butters with DAP were consistently lower than the control sample throughout the storage period, with significant differences observed (P< 0.05) (Table 3). Increasing DAP concentration resulted in decreased pH values, with the lowest pH recorded in the sample containing 25% DAP at the end of storage. This decrease is attributed to the low pH of DAP (4.18 ± 0.01) . Titration Acidity (LA%) values increased during storage, with the control sample showing lower acidity compared to the other samples. Higher concentrations of DAP led to increased acidity, with all samples significantly different (P< 0.05).

Free fatty acid (FFA) values in butters fortified with DAP were higher than in the control sample, varying with DAP concentration (P< 0.05). Significant differences in FFA values were observed over the storage period $(P< 0.05)$. This increase can be attributed to the low pH, high acidity (Table 3), and high phenolic and antioxidant activity of both DAP and the butters (Table 4). These results align with those of Shakerardekani et al. (2020), who reported that adding honey to pistachio butter increased FFA levels depending on the honey concentration. Importantly, the increased FFA content in DAP-fortified butters did not negatively impact sensory scores related to rancidity, as there is not always a positive correlation between FFA levels and sensory bitterness. Mahmoudi et al. (2019) found that adding Ziziphora clinopodioides to butter did not change acidity values during a 10-day storage period. Carotenoid content significantly affects the color of dairy products, especially high-fat products like butter, by imparting a yellow hue (O'Callaghan et al., 2016). This color characteristic is crucial for consumer acceptance.

The color parameters (L^*, a^*, b^*) are presented in Table 3. The brightness values

 $Samples^b$ Dry-matter Fat Fructose Glucose Total sugar A0 76.87±0.47^a 73.75±0.96a - - - A₁₅ 73.42 ± 0.45^b 65.25 ± 0.50^b 1.33 ± 0.03^a
A₂₀ 72.73 ± 0.68^b 65.50 ± 0.58^b 1.69 ± 0.03^b 2.36 ± 0.04^a 3.69 ± 0.06^a A_{15} 73.42 ± 0.45^b 65.25 ± 0.50^b 1.33 ± 0.03^a 2.36 ± 0.04^a 3.69 ± 0.06^a
 72.73 ± 0.68^b 65.50 ± 0.58^b 1.69 ± 0.03^b 3.01 ± 0.05^b 4.70 ± 0.06^b 65.50 ± 0.58^b 1.69 ± 0.03^b 3.01 ± 0.05^b 4.70 ± 0.06^b 4.70 ± 0.06^b A₂₅ $71.47 \pm 0.57^{\circ}$ $65.50 \pm 0.58^{\circ}$ $2.03 \pm 0.03^{\circ}$ $3.61 \pm 0.06^{\circ}$ 5.64 2.03 ± 0.03 ^c 3.61 ± 0.06 ^c 5.64 ± 0.08 ^c

Table 2. Proximate composition $(\%)$ of control and flavored butters.^{*a*}

 a (a-c): The data are presented as means \pm standard deviation. Significant differences among the means within the same column and row are denoted by distinct lowercase and uppercase superscript letters, respectively, at a significance level of $P < 0.05$.

 \overrightarrow{A}_0 : Control butter without DAP, A₁₅: Butter containing 15% DAP, A₂₀: Butter containing 20% DAP, A₂₅: Butter containing 25% DAP.

	Storage			$Samples^b$	
Properties	(Days)	A_0	A_{15}	A_{20}	A_{25}
		4.65 ± 0.11^{Aa}	4.26 ± 0.01^{Bb}	4.24 ± 0.01^{Bb}	4.21 ± 0.01^{Bb}
	15	4.55 ± 0.01 ^{Ab}	4.27 ± 0.01 ^{Bb}	4.21 ± 0.01 ^{Cd}	4.17 ± 0.01 ^{Dd}
pH	30	4.66 ± 0.01 Aa	4.38 ± 0.01 ^{Ba}	4.35 ± 0.01 ^{Ca}	4.27 ± 0.01 ^{Da}
	45	4.53 ± 0.01 ^{Ab}	$4.18{\pm}0.01^{\text{ Bd}}$	4.11 ± 0.01 $^{\rm Ce}$	4.10 ± 0.01 ^{Ce}
	60	4.59 ± 0.01 Aab	4.24 ± 0.01 ^{Bc}	4.22 ± 0.01 ^{Cc}	4.18 ± 0.01 ^{Dc}
	$\mathbf{1}$	0.20 ± 0.02 ^{Cb}	0.35 ± 0.01 ^{Bb}	0.37 ± 0.01^{Bb}	0.44 ± 0.01^{Ac}
Titratable	15	$0.19 \!\!\pm\!\! 001^{\,\mathrm{Db}}$	0.37 ± 0.01 ^{Bab}	0.38 ± 0.01 ^{Bb}	$0.44{\pm}0.01$ $^{\rm Ac}$
Acidity	30	$0.17{\pm}0.01^{\text{ Dc}}$	0.33 ± 0.01 ^{Cc}	$0.38{\pm}0.01^{\ Bb}$	$0.45{\pm}0.01$ $^{\rm Abc}$
$(LA\%)$	45	$0.19 \pm 0.01^\text{Db}$	0.37 ± 0.01 ^{Ca}	$0.41{\pm}0.01$ $^{\mathrm{Ba}}$	0.47 ± 0.01 $^{\rm{Aa}}$
	60	0.23 ± 0.01 ^{Da}	$0.38{\pm}0.01^{\mathrm{Ca}}$	$0.42{\pm}0.01$ $^{\mathrm{Ba}}$	0.46 ± 0.01 $^{\mathrm{Ab}}$
	$\mathbf{1}$	1.32 ± 0.06^{De}	3.14 ± 0.13 ^{Cc}	3.53 ± 0.07 ^{Bab}	3.87 ± 0.07 Aa
FFA	15	1.76 ± 0.06 ^{Dc}	3.20 ± 0.06 Cbc	3.40 ± 0.06 ^{Bc}	3.93 ± 0.13 Aa
$(mg KOH g^{-1})$	30	$1.60 \!\!\pm\!\! 0.11^{\,\mathrm{Dd}}$	$2.10{\pm}0.06^{\,\mathrm{Cd}}$	2.27 ± 0.06 $^{\mathrm{Bd}}$	$2.44{\pm}0.06$ $^{\rm Ac}$
butter)	45	$2.44{\pm}0.11^{\,\mathrm{Da}}$	$3.28{\pm}0.06^{\text{Cb}}$	3.45 ± 0.06 ^{Bbc}	3.62 ± 0.06 ^{Ab}
	60	1.99 ± 0.06 ^{Db}	3.45 ± 0.06 ^{Ca}	3.56 ± 0.06 ^{Ba}	$3.81{\pm}0.09$ $^{\rm{Aa}}$
	$\mathbf{1}$	88.00 ± 0.43 ^{Ac}	84.46 ± 0.24^{Bb}	$85.\overline{14\pm0.45^{Ba}}$	81.55 ± 0.67 ^{Cab}
	15	89.75 ± 0.18 ^{Ab}	84.70 ± 0.14 ^{Bb}	82.71 ± 0.58 ^{Cb}	80.63 ± 0.19 ^{Dbc}
L^*	30	87.15 ± 0.51 ^{Ad}	80.95 ± 0.74 ^{Cd}	83.33 ± 0.42 ^{Bb}	78.81 ± 0.49 ^{Dd}
	45	90.56 ± 0.37 $^{\rm Aa}$	85.75 ± 0.52 ^{Ba}	85.03 ± 0.59 ^{Ba}	$81.77{\pm}0.98^{\,\mathrm{Ca}}$
	60	88.21 ± 0.52 ^{Ac}	82.39 ± 0.33 ^{Bc}	80.43 ± 0.32 ^{Cc}	79.79 ± 0.91 ^{Ccd}
	$\mathbf{1}$	-3.84 ± 0.07 ^{Bb}	-3.72 ± 0.31 ^{Ba}	-3.56 ± 0.34 ^{Bb}	-2.40 ± 0.71^{Aa}
	15	-3.92 ± 0.05 ^{Bb}	-3.23 ± 0.29 ^{Aa}	-2.88 ± 0.33 ^{Aa}	-2.83 \pm 0.30 ^{Aab}
a^*	30	-3.85 ± 0.11 ^{Ab}	-3.70 ± 0.14 ^{Aa}	-4.09 ± 0.38 Ac	-4.02 ± 0.37 ^{Ad}
	45	-3.48 ± 0.12 Aa	-3.60 ± 0.56 $^{\rm{Aa}}$	-3.65 ± 0.03 ^{Ab}	-3.83 ± 0.19 Acd
	60	-3.51 ± 0.17 Aa	-3.61 ± 0.10 ^{Aa}	-3.50 ± 0.19 ^{Ab}	-3.27 ± 0.55 Abc
	$\mathbf{1}$	17.02 ± 0.83 ^{Cb}	23.07 ± 0.71 ^{Bb}	22.90 ± 1.04 ^{Bab}	30.39 ± 0.93 ^{Aa}
	15	17.11 ± 0.29 ^{Cb}	23.96 ± 0.77 ^{Bb}	$23.18{\pm}0.86$ $^\text{Bab}$	$29.06{\pm}0.96$ $^{\mathrm{Aab}}$
b^*	30	$16.59{\pm}0.60^{\text{Cb}}$	23.66 ± 0.73 ^{Bb}	22.60 ± 1.55 ^{Bb}	28.33 ± 1.11 ^{Ab}
	45	16.50 ± 0.16^{Db}	$26.50 \pm 1.04^{\, \text{Ba}}$	$24.63{\pm}1.36^{\,\mathrm{Ca}}$	29.05 ± 0.73 Aab
	60	18.17 ± 0.50 ^{Da}	21.23 ± 0.59 ^{Cc}	24.48 ± 0.99 ^{Bab}	26.37 ± 0.70 Ac

Table 3. Evolution of physicochemical properties of the control and flavored butters during storage.⁴

^a(A-D and a-d): The data are presented as means±standard deviation. Significant differences among the means within the same column and row are denoted by distinct lowercase and uppercase superscript letters, respectively, at a significance level of P< 0.05.
^b A₀: Control butter without DAP, A₁₅: Butter containing 15% DAP, A₂₀: Butter containing 20% DAP, A₂₅: Butter

containing 25% DAP

(L*) exhibited a significant decrease in butters containing DAP ($P < 0.05$), while the redness (a^*) and yellowness (b^*) values increased with higher concentrations of DAP $(P< 0.05)$. Specifically, the lowest L^{*} value was observed in the 25% DAP sample (81.55 ± 0.67) , whereas the highest L* value was recorded in the control sample (88.00 ± 0.43) on the initial day of storage. This observation is likely attributed to the inherent color properties of DAP, with respective values of L^* (48.80 \pm 0.55), a* (3.70 ± 0.37) , and b* (37.20 ± 0.56) . Notably, there were no significant alterations in the L^* value of the control sample over time (P> 0.05), whereas a decrease was noted in $(P<$ DAP-fortified samples 0.05). Conversely, the b* values demonstrated an increase in DAP-containing butters, significantly differing from the control sample ($P < 0.05$), with the highest b* values consistently observed in the 25% DAP throughout This sample storage. phenomenon can be attributed to the elevated phenolic content in apricot, particularly beta-carotene, contributing to a more pronounced yellowish hue. Contrastingly, O'Callaghan et al. (2016) documented a reduction in b* values in butters following six months of storage, suggesting a transition towards a paler blue tint at 4 °C. These findings diverge from those of Göksel et al. (2016), who observed an augmentation in all color parameters of butters upon the addition of orange dietary fiber concentrate and stone pear dietary fiber concentrate.

PV Values of Butters

PV (meq O_2 kg⁻¹ butter) results for butters with DAP and the control are presented in Figure 1. PV values are crucial for assessing butter oxidation. Although traditional PV determination methods indicate butter quality, they do not distinguish between various unsaturated fatty acids undergoing oxidation, serving only as a primary oxidation indicator (Samet-Bali et al., 2009).

As shown in Figure 1, no PV formation was detected on day 1 in the control sample, while it was observed on day 15 in butters with DAP. The absence of peroxide formation during the first 15 days in DAPcontaining butters suggests that no autooxidation occurred, with a significant increase in PV indicating the onset of autooxidation (Farag et al., 1990). Higher DAP concentrations resulted in decreased PV values (P< 0.05), likely due to different fat and phenolic contents in the butters. Incorporating 25% DAP inhibited peroxide formation during 60 days of storage, consistent with Asha et al. (2015), who found that adding orange peel extract lowered PV in butter samples during storage. Conversely, Zhao and Hall (2007) reported that high concentrations of dried grape extracts increased PV in butter model systems, attributing this to high ferrous content in grapes, which increases lipid oxidation. They also noted that dried grape extracts did not prevent free radical formation in butter model systems with sunflower oil due to restricted phenolic mobility and migration. Nadeem et al. (2014) found that adding sesame extract to butter with olein increased PV throughout storage. Some researchers observed higher lipid peroxidation in samples with higher concentrations of antioxidant extracts from tomatoes, due to the prooxidant properties of bioactive components (Abid et al., 2017). Thus, factors such as food composition, processing conditions, extract amount, and storage significantly impact antioxidant properties. In delaying oxidation product formation, the effective DAP ratios were 15, 20, and 25% for 30, 45, and 60 days of storage, respectively. During peroxidation, radical species degrade fatty acids and other

Figure 1. Evolution of PV values of control and flavored butters during storage. Data are means \pm standard deviation; different lowercase and uppercase superscript letters for storage and samples, respectively, indicate that the means differ significantly ($P < 0.05$).

lipid components, such as carotenoids, chlorophyll pigments, and tocopherols (Hornero-Méndez et al., 2001). The same authors reported that the lipid fraction in paprika is rich in polyunsaturated fatty acids, and the presence of carotenoids vulnerable to peroxidation could explain the higher PV increase in oleoresins. Mahmoudi et al. Ziziphora clinopodioides (2019) used essential oil at rates of 300, 600, and 900 ppm, finding that PV in butters increased at 600 ppm but decreased at 900 ppm during a 10-day storage period.

TPC, DPPH Radical Scavenging Activity and TAC of Butters

Natural antioxidants exert their effects through various mechanisms, such as preventing chain initiation by scavenging radicals, chelating metals, reducing localized oxygen concentrations, and decomposing peroxides (Baiano et al., 2009). Table 4 presents the Total Phenolic Content (TPC), DPPH scavenging activity, and Total

Antioxidant Capacity (TAC) of the control butters and those enriched with DAP.

Significant differences $(P< 0.05)$ were observed between the phenolic contents of the control and DAP-containing butter samples. The high phenolic content of apricot pulp (1592.08±57.459 mg GAE kg⁻¹ butter) accounts for these differences. As the DAP ratio increased, the phenolic content also increased, in the order of $A_{25} > A_{20} >$ A_{15} A₀ on the first day of storage. However, the TPC of butters was not as high as that of apricot pulp itself, likely due to the phenolic differential distribution of substances within the butter matrix. This result aligns with Vidanagamage et al. (2016) , who found that adding cinnamon extract to butter yielded lower phenolic content compared to the extract itself. Phenolic contents of all butter samples increased during storage, potentially due to the Folin-Ciocalteu method, which can be influenced by reducing sugars, aromatic amines, sulfur dioxide, ascorbic acid, organic acids, and other compounds (Castro-Lopez et al., 2016). Nicoli et al. (1997) also

Table 4. Variations in TPC, DPPH, and TAC of the control and flavored butters during storage.⁴

Properties	Days			Samples ^b	
		A_0	A_{15}	A_{20}	A_{25}
		159.27 ± 5.74 ^{Dd}	392.09 ± 4.96 ^{Cd}	420.21 ± 7.50^{Bb}	454.58 ± 3.61 ^{Ad}
TPC	15	282.19 ± 7.30^{Dc}	416.05 \pm 3.80 $\mathrm{^{Cc}}$	$436.88 {\pm} 2.95$ Bbc	$488.44{\scriptstyle\pm4.92}$ $^{\rm Ac}$
$(mg \text{ GAE kg}^{-1})$	30	191.04 ± 5.89^{Dd}	324.90 ± 1.04 $^{\mathrm{Ce}}$	416.04 ± 7.22 ^{Bc}	495.21 \pm 2.95 ^{Ac}
butter)	45	518.13 ± 9.62^{Db}	931.67 \pm 7.12 ^{Cb}	1016.56 ± 6.45 $^{\text{Ba}}$	1062.40 ± 8.74 ^{Aa}
	60	934.80 ± 2.41^{Da}	944.69 ± 3.56 Ba	1035.83 ± 2.08 ^{Ba}	1054.59 ± 2.69 ^{Ab}
		23.12 ± 0.96 ^{Bbc}	25.94 ± 0.42^{Ab}	26.25 ± 0.20^{Ac}	26.70 ± 0.10^{Ac}
DPPH	15	22.36 ± 0.17 ^{Dc}	25.71 ± 0.16 ^{Cb}	26.80 ± 0.16 $^{\mathrm{Bb}}$	28.99 ± 0.12 ^{Ab}
(Inhibition %)	30 45	$25.20{\pm}1.08^{\text{ Ca}}$	$27.41{\scriptstyle \pm 0.30\,}^{\rm Ba}$	$30.42{\pm}0.37^{\,\rm{Aa}}$	$30.04 {\pm} 0.28$ $^{\mathrm{Aa}}$
		22.23 ± 0.48 ^{Cc}	$26.02{\pm}0.32$ $^{\mathrm{Bb}}$	$26.48 {\pm} 0.10^{\textrm{\,ABbc}}$	26.82 ± 0.17 ^{Ac}
	60	23.73 ± 0.29 ^{Db}	24.82 ± 0.15 ^{Cc}	25.37 ± 0.19 ^{Bd}	26.64 ± 0.07 Ac
		73.78 ± 1.38 ^{Db}	$113.1 \overline{1 \pm 081}^{\text{Ce}}$	127.56 ± 1.38 ^{Be}	135.89 ± 0.99 ^{Ae}
$_{\mathrm{TAC}}$	15	$81.89 {\pm} 1.68$ $^{\text{Da}}$	126.89 ± 0.63 ^{Cd}	173.56 ± 4.65 ^{Bd}	185.56 ± 0.81 ^{Ad}
(mg)	30	72.11 ± 1.22^{Db}	165.11 ± 1.31 ^{Cc}	$188.89{\pm}0.57$ Bc	194.89 \pm 2.97 ^{Ac}
AAE kg ⁻¹)	45	$67.22{\pm}0.92$ $^{\rm Dc}$	$174.78{\pm}1.90$ ^{Cb}	193.44 ± 2.60 ^{Bb}	206.33 ± 1.68 ^{Ab}
	60	$63.89 {\pm} 0.76^{\,\textrm{Dd}}$	248.45 ± 1.18 ^{Ca}	$259.67 {\pm} 1.83$ $^{\text{Ba}}$	271.22 ± 1.68 Aa

 a (A-D and a-e): The data are presented as means± standard deviation. Significant differences among the means within the same column and row are denoted by distinct lowercase and uppercase superscript letters, respectively, at a significance level of $P < 0.05$.

 b A₀: Control butter without DAP, A₁₅: Butter containing 15% DAP, A₂₀: Butter containing 20% DAP, A₂₅: Butter containing 25% DAP

Samples ^{b}	Storage	Colour	Texture	Odor	Taste	Rancidity	General
	(Days)						Acceptability
		3.57 ± 0.20^{aD}		3.93 ± 0.10^{D} 3.86 ± 0.40^{B}	3.86 ± 0.21 ^C	$5.86 \pm 1.21^{\mathrm{B}}$	$5.57 \pm 0.61^{\rm B}$
	15	$3.57 \pm 0.00^{\mathrm{aD}}$		$4.08 \pm 0.30^{\circ}$ 3.93 $\pm 0.51^{\circ}$	3.65 ± 0.30 ^C	$5.79 \pm 1.11^{\rm B}$	$5.50 \pm 0.51^{\rm B}$
	30	$3.43 \pm 0.00a^C$		$3.93 \pm 0.30^{\circ}$ 3.72 $\pm 0.40^{\circ}$	$3.36 \pm 0.30^{\rm B}$	$5.79 \pm 1.11^{\text{A}}$	$5.36 \pm 0.71^{\rm B}$
A_0	45	$3.43 \pm 0.00a^C$		$3.71 \pm 0.00^{\circ}$ 3.64 $\pm 0.71^{\circ}$	$3.36 \pm 0.30^{\rm B}$	$5.58 \pm 1.01^{\rm B}$	$5.22 \pm 0.50^{\rm B}$
	60	$3.36 \pm 0.30a^C$		$3.57 \pm 0.00^{\circ}$ 3.58 $\pm 0.40^{\circ}$	$3.43 \pm 0.40^{\rm B}$	$5.65 \pm 1.11^{\mathrm{B}}$	$5.15 \pm 0.40^{\circ}$
		6.22 ± 0.11 ^C		$6.36 \pm 0.30^{\circ}$ 7.79 \pm 0.11a ^A	7.36 ± 0.71 ^B	$7.61 \pm 0.14^{\rm A}$	7.58 ± 0.40 ^A
	15	6.29 ± 0.21 ^C		$6.29 \pm 0.00^{\mathrm{B}}$ 7.57 \pm 0.00 ^{bA}	7.36 ± 0.31 ^B	$7.43{\scriptstyle \pm0.40^{\mathrm{AB}}}$	$7.72 \pm 0.60^{\rm B}$
	30	$6.50 \pm 0.10^{\rm B}$		$6.07 \pm 0.71^{\rm B}$ 7.36 \pm 0.10 ^{bA}	$7.29 \pm 0.40^{\rm A}$	7.07 ± 0.51 ^A	7.65 ± 0.30 ^A
A_{15}	45	$6.36 \pm 0.30^{\rm B}$		$6.07 \pm 0.71^{\rm B}$ 7.36 \pm 0.10 ^{bA}	7.22 ± 0.50 ^A	$7.00 \pm 0.41^\mathrm{AB}$	$7.43 \pm 0.61^{\rm A}$
	60	$6.43 \pm 0.20^{\rm B}$		$5.93 \pm 0.71^{\mathrm{B}}$ 7.43 $\pm 0.00^{\mathrm{bAB}}$	7.07 ± 0.51 ^A	$6.86 {\pm} 0.40^{\mathrm{AB}}$	7.15 ± 0.21 ^B
		$8.29 \pm 0.00^{\text{ B}}$		$7.36 \pm 0.30^{\mathrm{B}}$ 8.00 \pm 0.20 ^A	8.65 ± 0.50 ^{aA}	$8.79 \pm 0.11^{\text{A}}$	8.43 ± 0.20^{aA}
	15	$8.29 \pm 0.21^{\mathrm{B}}$		$7.50\pm0.30^{\rm A}$ 8.29 $\pm0.00^{\rm A}$	8.50 ± 0.51 ^{aA}	$8.57 \pm 0.20^{\rm A}$	8.36 ± 0.10 abB
	30	$8.36 \pm 0.30^{\rm A}$		7.57 ± 0.20 ^A 8.22 ± 0.50 ^A	7.36 ± 0.10^{bA}	$8.36 \pm 0.71^{\text{A}}$	8.22 ± 0.11^{abA}
A_{20}	45	$8.22 \pm 0.30^{\rm A}$		$7.50\pm0.30^{\rm A}$ 8.07 \pm 0.51 ^A	7.29 ± 0.21 ^{bA}	8.22 ± 0.91 ^A	8.07 ± 0.10^{bA}
	60	$8.07 \pm 0.10^{\rm A}$		$7.36 \pm 0.30^{\rm A}$ $7.93 \pm 0.30^{\rm A}$	7.36 ± 0.30 ^{bA}	8.07 ± 0.71 ^A	8.07 ± 0.10^{bA}
		8.86 ± 0.00 ^{aD}		8.07 \pm 0.10 ^A 7.64 \pm 0.10 ^{aA}	7.36 ± 0.10^{abB}	$8.86 \pm 0.00^{\rm A}$	8.07 ± 0.51 ^A
	15	8.79 ± 0.11 ^{aA}		$8.15 \pm 0.21^{\text{A}}$ 7.57 $\pm 0.20^{\text{abA}}$	$7.43{\pm}0.00^{\mathrm{a\,B}}$	8.65 ± 0.30 ^A	$8.15 \pm 0.40^{\rm B}$
	30	8.64 ± 0.10^{abA}		8.08 ± 0.30 ^A 7.43 ± 0.20 ^{abA}	7.36 ± 0.10^{abA} 7.86 ± 1.41^{A}		7.93 ± 0.30 ^A
A_{25}	45	$8.50 \pm 0.10^{b c A}$		$7.93 \pm 0.30^{\rm A}$ $7.00 \pm 0.41^{\rm bA}$		7.07 ± 0.10^{bA} 7.43 $\pm1.01^{AB}$	$7.79 \pm 0.11^{\rm A}$
	60	8.36 ± 0.10 ^{cA}		$7.72 \pm 0.40^{\mathrm{A}}$ $7.00 \pm 0.00^{\mathrm{bB}}$		7.15 ± 0.21^{abA} 7.14 ± 0.61^{AB}	7.64 ± 0.10 ^{AB}

Table 5. Variations in Sensory properties of control and flavored butters during storage.⁴

 a (A-D and a-c): The data are presented as means \pm standard deviation. Significant differences among the means within the same column and row are denoted by distinct lowercase and uppercase superscript letters, respectively, at a significance level of $P < 0.05$.

 b A₀: Control butter without DAP, A₁₅: Butter containing 15% DAP, A₂₀: Butter containing 20% DAP, A₂₅: Butter containing 25% DAP.

noted that Maillard reaction products, formed during the heating of apricot, could enhance the overall antioxidant properties by minimizing the loss of natural antioxidants.

DPPH scavenging activity was higher in DAP-fortified butters than in the control (Table 4). DPPH values ranged from 22.23±0.48 to 25.20±1.08% in the control sample and from 24.82 \pm 0.48 to 30.42±0.48% in butters produced with DAP during storage. This can be attributed to the DPPH activity of the pulp $(40.57\pm0.00\%)$ and the heating process for pasteurization, which can generate non-nutrient antioxidant components (Nicoli et al., 1997). DPPH values increased with higher apricot butter ratios (P< 0.05), though a significant decrease was observed at the end of storage compared to the first day in DAP-fortified butters ($P < 0.05$).

The TAC results of the samples are shown in Table 4, and varied with the addition of pulp $(P< 0.05)$, with values ranging from 63.89±0.76 to 73.78±1.38 mg AAE kg^{-1} butter for the control sample (A_0) , 113.11±0.81 to 248.45±1.18 mg AAE kg^{-1} for A_{15} , 127.56±1.38 to 259.67±1.83 mg AAE kg⁻¹ Gallicfor for A₂₀, and 135.89±0.99 to 271.22±1.68 mg AAE kg⁻¹ butter for A₂₅. Storage butter for A_{25} . Storage significantly affected the TAC values of all samples (P< 0.05), with TAC decreasing in the control sample while increasing in butters with DAP during storage. This can be attributed to the high antioxidant (639.11 \pm 0.94 mg AAE kg⁻¹) and phenolic content of apricot pulp. Additionally, there was a strong positive correlation between antioxidant capacity and phenolic content during storage (Martinez-Flores et al., 2015).The increase in phenolic content of butters with DAP during storage can also be linked to the 28% total sugar content, phenolic content, DPPH activity, and TAC of apricot pulp. The interactions between phenolic

compounds and components of spices, herbs, and fruits, as well as their partitioning between aqueous and oily fractions, also play a role (Baiano et al., 2009). Kopjar et al. (2016) noted that the addition of sugars such as sucrose and trehalose to the model systems containing phenolics like catechin, quercetin, and gallic acid influenced antioxidant activity through a synergistic effect, where sugars can preserve or even enhance polyphenol activity (Peinado et al., 2010).

Sensory Evaluation

The primary factors influencing consumer butter purchases are color and taste. The sensory properties of the butter samples, detailed in Table 5, revealed significant differences in color and texture among the samples ($P < 0.05$). The incorporation of DAP notably enhanced the color scores of the butters, with DAP-fortified samples receiving higher preference from panelists compared to the control. As the concentration of DAP increased, the color scores improved correspondingly ($P \leq 0.05$), with the highest scores observed in the sequence $A_{25} > A_{20} > A_{15} > A_0$. Notably, the storage period did not significantly impact the color and texture of the butter samples $(P> 0.05)$.

The addition of apricot pulp also had a significant effect on odor values compared to the control sample $(P< 0.05)$. Taste scores were consistently higher in DAP-fortified butters than in the control, with the peak score recorded for sample A_{20} up to day 15. After this period, no significant differences were observed among the apricot butter samples.

The improvement in taste over time may be attributed to the gradual diffusion of flavor compounds from the apricot into the butter. Furthermore, rancidity and overall acceptability scores were higher for the DAPfortified butters, demonstrating significant differences from the control ($P < 0.05$). These findings suggest that the addition of apricot pulp, irrespective of the ratio, enhances

sensory properties and is preferred by panelists.

CONCLUSIONS

This study aimed to improve the functional properties and oxidative stability of butter while introducing an alternative variety to other dairy products by incorporating Dried Apricot Pulp (DAP), known for its significant phenolic content and antioxidant activity, without compromising sensory perception. The results of this study demonstrated an increase in phenolic content with the addition of DAP. Moreover, it was observed that the addition of 25% DAP effectively reduced peroxide values, indicating a decrease in primary oxidation. Butters fortified with DAP exhibited higher sensory scores, particularly in terms of taste and color. All concentrations of DAP (15, 20, and 25%) were found to be acceptable, with the 25% concentration being the most preferred by panelists. Butter is a staple dairy product commonly consumed with jam and honey during breakfast. However, the high sugar content in jam and honey may limit its consumption for individuals with weight management issues or conditions such as diabetes. The addition of 25% DAP, which contains only 5.64±0.08% sugar, was sufficient to enhance the taste and sweetness of butter, making it a viable alternative to traditional spreads. Furthermore, the lower fat content in butter fortified with DAP may appeal to individuals seeking healthier food options with functional properties. Overall, the sensory evaluation revealed that butter fortified with DAP received favorable scores compared to regular butter. This suggests that the production of butter with DAP not only enhances its antioxidant properties but also improves sensory perception. Importantly, unlike previous studies, no sensory limitations were identified with the addition of DAP.

Future research should focus on optimizing production processes for commercial and industrial-scale applications of DAP.

ACKNOWLEDGEMENTS

This paper is based on the thesis entitled "The Determination of Some Quality Properties of Butters Produced with Dried Apricot" by Merve Bulut. We thank Assoc. Prof. Dr. Cemalettin Baltacı for sugar analyses, Prof. Dr. Necati ÇELİK for checking the language.

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ظرفیت آنتی اکسیدانی، پایداری اکسیداتیو و خواص حسی کره طعمدار با استفاده از خمیر (pulp) زردالوی خشک

م. بولوت، و ا. گوندوغدو

چکیده

در این پژوهش، غلظتهای مختلف خمیر زردآلوی خشک (DAP) در سطوح ۰، ۱۵، ۲۰ و ۲۵ درصد به کره اضافه شد و اثرات آن در مدت ۴۵ روز انبارداری (storage (بر ظرفیت آنتیاکسیدانی، پایداری اکسیداتیو، رنگ و ویژگیهای حسی مورد بررسی قرار گرفت. نمونههای کره بهعنوان A₀ (شاهد)، A₁₅. و A25 مطابق با سطوح مربوطه اختلاط DAP برچسبگذاری شد. افزودن DAP منجر به کاهش $\rm A_{20}$ قابل توجهی در pH شد، در حالی که اسیدیته قابل تیتراسیون (بیان شده به عنوان درصد اسید لاکتیک) و محتوای اسیدهای چرب آزاد (FFA) (میلی گرم KOH در هر گرم کره) به صورت وابسته به دوز(–dose dependent)، افزایش یافت (P۰.۰۵ >). غنی سازی با DAP تشکیل پراکسید را در نمونه های کره به تاخیر انداخت. افزون بر این، مکمل DAP به طور قابل توجهی (۰.۰۵P (< محتوای فنلی کل (TPC (را افزایش داد. فعالیت مهاری (2)،2-دی فنیل-۱-پیکریل هیدرازیل) DPPH را بهبود بخشید و ظرفیت آنتی اکسیدانی کل (TAC (را در طول انبارداری افزایش داد. اعضای پانل نمرات حسی بالاتری را به کره های غنی شده با DAP اختصاص دادند. این کره های غنی شده محتوای چربی کمتری (از ٪۶۵.۲۵ تا ٪۶۵.۵۰) در مقایسه با کره شاهد (٪۷۳.۷۵) نشان دادند. شایان ذکر است میزان کل قند

نمونه های A₂₅ و A₂₅ به ترتیب ۰۶/۰۱:۰۶۹)۳، ۰/۰۷-۰۷۷۰ و ۵/۶۴±۰/۰۸ بود. به طور کلی، این بررسی پتانسیل همه نسبتهای DAP، بهویژه ۲۵ درصد را، به عنوان منابع غنی آنتیاکسیدانها نشان میدهد. با این حال، تنظیمات بیشتر فرمول و تجزیه و تحلیل های جامع برای کاربردهای صنعتی و بازاریابی ضروری است.